

Pondaplin: A Novel Cyclic Prenylated Phenylpropanoid from *Annona glabra*

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Abstract: A novel cyclic prenylated phenylpropanoid, pondaplin was isolated from the ethanolic extracts of the leaves of *Annona glabra*, by directing the fractionation with the brine shrimp lethality test (BST). Pondaplin showed selective cytotoxicities at moderate potencies among six human solid tumor cell lines. © 1998 Published by Elsevier Science Ltd. All rights reserved.

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Annona glabra L. (Annonaceae), commonly known as pond apple, is a tropical tree distributed mainly in the Americas and in southeast Asia. It is used in traditional medicine as an insecticide and a parasiticide. Several bioactive Annonaceous acetogenins have been previously isolated from this species. As part of our continuing efforts to find new and structurally diverse bioactive leads, a novel cyclic prenylated phenylpropanoid, pondaplin (1), was isolated from the bioactive ethanolic extracts of the leaves, obtained from trees native to Florida, using bioactivity-directed fractionation with the brine shrimp lethality test (BST). The structure of 1 (Figure 1) was identified as a cyclic prenylated p-coumarate by NMR spectroscopic techniques (Table 1, Figure 2). The new compound demonstrated moderate cytotoxicities among six human solid tumor cell lines with selectivities for the breast (MCF-7) and prostate (PC-3) cancer cell lines (Table 2).

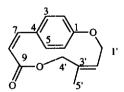


Figure 1. Structure of Pondaplin (1).

Compound 1 was isolated as colorless crystals, m. p. 194-195 °C. Its molecular weight was suggested by the peak at m/z 231 [MH]⁺ in the CIMS. The HRCIMS gave m/z 231.1018 for the [MH]⁺ ion (calcd. 231.1021) corresponding to the molecular formula, $C_{14}H_{14}O_3$.

Compound 1 showed a UV λ_{max} at 245 nm (log ϵ = 3.05) with a shoulder at 295 nm (log ϵ = 3.03) and aromatic (3040, 1603, and 1513 cm⁻¹) and carbonyl (1709 cm⁻¹) absorption peaks in its IR spectrum. The presence of a *para*-substituted aromatic system was suggested by δ_H 7.58 (dd, J=9.0 Hz, 2.0 Hz, H-3/5) and δ_H 6.80 (dd, J=8.5 Hz, 2.0 Hz, H-2/6) in the NMR (Table 1). This was confirmed by the ¹³C chemical shifts at δ_C 157.99 (C-1), 114.94 (C-2/6), 132.97 (C-3/5), and 126.46 (C-4). The linkage of the double bond (C-7/8) to the aromatic system was indicated by single-relay COSY data. The linkage was confirmed because the olefinic doublets at δ_H 6.87 (H-7) and δ_H 5.82 (H-8) showed HMBC correlations to aromatic carbons at δ_C 132.97 (C-3/5) and to δ_C 126.46 (C-4), respectively (Figure 2).



Figure 2. Selected HMBC correlations in 1

The presence of an ester linkage in 1 was indicated by the carbonyl carbon signal at δ_C 166.59 together with IR absorption at 1709 cm⁻¹. The HMBC correlations between the signal of H-7 (δ_H 6.87) and the carbonyl group (δ_C 166.59) suggested that the carbonyl group must be located at C-9.

Table 1. ¹³ C NMR and ¹ H NMR (δ , J in Hz) of 1.				
	¹³ C NMR (125 MHz)	¹ H (500 MHz) (<i>J</i> in Hz)		
1	157.99	•		
2	114.94	6.80 (dd, 9.0, 2.0)		
3	132.97	7.58 (dd, 8.5, 2.0)		
4	126.46	-		
5	132.97	7.58 (d, 9.0)		
6	114.94	6.80 (d, 9.0)		
7	144.21	6.87 (d, 12.0)		
8	116,06	5.82 (d, 12.5)		
9	166.59	-		
1,	58.53	4.17 (d, 7.0)		
2'	126.97	5.62 (td, 6.5, 1.5)		
3'	132.19	<u>.</u>		
4'	68.77	4.54 (s)		
5'	13.83	1.67 (s)		

The remaining four resonances at δ_H 5.62 (1H, td, J=6.5 Hz, 1.5, H-2'), 4.54 (2H, s, H-4'), 4.17 (2H, d, J=7.0 Hz, H-1'), and 1.67 (3H, s, H-5') and five peaks at δ_C 132.19 (C-3'), 126.97 (C-2'), 68.77 (C-4'), 58.53 (C-1'), and 13.83 (C-5') in the NMR spectra (Table 1) were characteristic spectral features for the oxygen-substituted prenyl group, which was

confirmed by HMBC correlations among H-1'/C-3', H-4'/C-2'. H-4'/C-3', H-5'/C-2', H-5'/C-3', and H-5'/H-4' (Figure 2).

The linkage between the ester group and the prenyl group was established via HMBC correlation between the methylene group (H-4') and the carbonyl carbon (C-9). The presence of an additional downfield-shifted methylene carbon signal at δ_C 58.53 (C-1') could only be accounted for by its connection to an oxygen atom. This was also revealed by the ¹H shifts of H-2/6 (δ_H 6.80) and the ¹³C NMR shift of C-1 (δ_C 157.99) in the aromatic system.

The coupling constant of the pair of olefinic protons at $\delta_{\rm H}$ 6.87 (H-7, br d, J =12.0 Hz) and 5.82 (H-8, br d, J=12.5 Hz) indicated the (Z)-configuration for the double bond at C-7/8. The stereochemistry of the other double bond (C-2'/3') was suggested by NOESY data. In the NOESY spectrum, the triple doublet signal of olefinic protons at $\delta_{\rm H}$ 5.62 showed a cross peak to the methyl group ($\delta_{\rm H}$ 1.67, s, 3H, H-5'), which indicated the (Z)-configuration.

Many of phenylpropanoids exhibit diverse biological activities of which the most noteworthy are antimicrobial, anticancer, and hypotensive properties. ⁸⁻¹⁰ Moreover, phenylpropanoid derivatives are already known to inhibit some enzymes such as cAMP phosphodiesterase and prostaglandin synthetase. ^{11, 12} The biological activities of 1 against six solid tumor cell lines are summarized in Table 2. The compound was moderately and selectively active across the six human tumor cell lines in our seven-day MTT human solid tumor cytotoxicity tests. ¹³ Since this is a small molecule and total synthesis should not be difficult, structural modifications with possible enhancement of bioactivity seems reasonable.

Table 2. Bioactivity of 1.

	Compound	1	adriamycin ^g
	BST ¹ ED ₅₀ (μg/ml)	3.9×10^{1}	-
Human	A-549 ^a	1.1 x 10 ¹	1.5 x 10 ⁻³
Tumor	MCF-7 ^b	2.3	1.1 x 10 ⁻¹
Cell	HT-29 ^c	5.0	2.6 x 10 ⁻²
Lines	A-498 ^d	2.6×10^{1}	3.0 x 10 ⁻³
ED_{50}	PC-3 ^e	3.6	1.9 x 10 ⁻²
(µg/ml)	PACA-2 ^f	1.9 x 10 ¹	1.6 x 10 ⁻³

¹Brine shrimp lethality test; ⁶ ^a Human lung carcinoma; ^b Human breast carcinoma; ^c Human colon adenocarcinoma; ^d Human kidney carcinoma; ^e Human prostate adenocarcinoma; ^f Human pancreatic carcinoma; ^g positive control standard.

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 and identified by one of us (Dr. Elsa Pilarinou). A voucher specimen is deposited in the Pharmacognosy
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